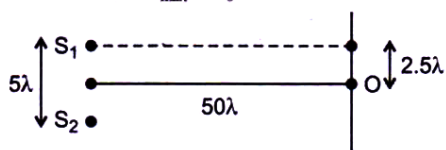


WEEKLY TEST OYM TEST - 20 BALLIWALA  
SOLUTION Date 01-09-2019

[PHYSICS]

1

We know that  $I_{\max} = I_0$



Given that  $d = 5\lambda$ , Hence  $2.5\lambda = \frac{d}{2}$

$$\text{Path diff.} = \frac{dy_n}{D} = \frac{d \times \frac{d}{2}}{10d} = \frac{d}{20} = \frac{\lambda}{4}$$

Phase diff. =  $90^\circ$

$$I = I_0 \cos^2 \frac{\phi}{2} = \frac{I_0}{2}$$

2.

If new value of distance of screen from double slit be  $D'$ , then

$$\beta' = \frac{\lambda D'}{d'} = \frac{\lambda D'}{(2d)} = \frac{\lambda D}{d} = \beta$$

or  $D' = 2D$ .

3.

For first minima,

$$a \sin \theta_1 = \lambda$$

$$\therefore a = \frac{\lambda}{\sin \theta_1} = \frac{6200 \times 10^{-10}}{\sin 30^\circ} = 1.24 \times 10^{-6} \text{ m}$$

$$= 1.24 \mu\text{m}.$$



4.

Suppose  $n_1$ th bright fringe of wavelength  $\lambda_1$  coincides with  $n_2$ th bright fringe of wavelength  $\lambda_2$ . Then,

$$\frac{n_1 \lambda_1 D}{d} = \frac{n_2 \lambda_2 D}{d}$$

or  $n_1 \lambda_1 = n_2 \lambda_2$   
 or  $\frac{n_1}{n_2} = \frac{\lambda_2}{\lambda_1} = \frac{10,000}{12,000} = \frac{5}{6}$

Let  $x$  be the given distance.

$$\therefore x = \frac{n_1 \lambda_1 D}{d}$$

Given that  $n_1 = 5, D = 2 \text{ m}$ ,

$$d = 2 \text{ mm} = 2 \times 10^{-3} \text{ m},$$

$$\lambda_1 = 12000 \text{ \AA} = 12000 \times 10^{-10} \text{ m} = 12 \times 10^{-7} \text{ m}$$

$$\therefore x = \frac{5 \times 12 \times 10^{-7} \times 2}{2 \times 10^{-3}} \\ = 6 \times 10^{-3} \text{ m} = 6 \text{ mm}$$

Hence, correct answer is (a).

5.

$$\text{Fringe width, } \beta = \frac{\lambda D}{d}$$

where  $\lambda$  is the wavelength of light,  $D$  is distance between slits and the screen,  $d$  is distance between the two slits.

As the  $D, d$  remain the same

$$\beta \propto \lambda$$

or  $\frac{\beta'}{\beta} = \frac{\lambda'}{\lambda}$

$$\text{or } \beta' = \frac{\lambda' \beta}{\lambda}$$

Substituting the given values, we get;

$$\beta' = \frac{4000 \text{ \AA} \times 3 \text{ mm}}{6000 \text{ \AA}} \\ = 2 \text{ mm.}$$

6.

$$\text{Fringe width, } \beta = \frac{\lambda D}{d}$$

$$\therefore D = \frac{\beta d}{\lambda} = \frac{4 \times 10^{-3} \times 0.1 \times 10^{-3}}{4 \times 10^{-7}} = 1 \text{ m.}$$

7.

$$\frac{I_{\max.}}{I_{\min.}} = \frac{(a_1 + a_2)^2}{(a_1 - a_2)^2} = \frac{(2 + 1)^2}{(2 - 1)^2} = \frac{9}{1}$$

8.

For maximum intensity on the screen,

$$d \sin \theta = n\lambda$$

$$\text{or } \sin \theta = \frac{n\lambda}{d} = \frac{(n)(2000)}{(7000)}$$

$$= \frac{n}{3.5}$$

Since,  $\sin \theta \leq 1$

$\therefore n = 0, 1, 2, 3$  only.

Thus, only seven maximas can be obtained on both sides of the screen.

9.

$$I = 4I_0 \cos^2 \left( \frac{\phi}{2} \right)$$

$$I_0 = 4I_0 \cos^2 \left( \frac{\phi}{2} \right)$$

$$\text{or } \cos \left( \frac{\phi}{2} \right) = \frac{1}{2}$$

$$\text{or } \frac{\phi}{2} = \frac{\pi}{3}$$

$$\text{or } \phi = \frac{2\pi}{3} = \frac{2\pi}{\lambda} \cdot \Delta x$$

$$\text{or } \frac{1}{3} = \frac{1}{\lambda} \cdot \left( \frac{yd}{D} \right)$$

$$\therefore y = \frac{\lambda}{3 \left( \frac{d}{D} \right)} = \frac{6 \times 10^{-7}}{3 \times 10^{-4}} = 2 \times 10^{-3} \text{ m} = 2 \text{ mm.}$$

10.

$$n_1 \lambda_1 = n_2 \lambda_2$$

$$n (7.8 \times 10^{-5}) = (n+1)(5.2 \times 10^{-5})$$

$$\text{or } n (2.6 \times 10^{-5}) = 5.2 \times 10^{-5}$$

$$\therefore n = 2.$$

11.

$$I = 2I_0(1 + \cos \delta)$$

When path difference =  $\lambda$ , then phase difference

$$\delta = \frac{2\pi}{\lambda} \times \text{path diff.} = 2\pi$$

$$\therefore I_1 = 2I_0(1 + \cos 2\pi) = 4I_0 = K \quad \dots(i)$$

When path difference =  $\lambda/4$ , then phase difference

$$\delta = \frac{2\pi}{\lambda} \times \frac{\lambda}{4} = \frac{\pi}{2}$$

$$\therefore I_2 = 2I_0 \left( 1 + \cos \frac{\pi}{2} \right) = 2I_0 = \frac{K}{2}$$

12.

13.

$$7\beta_1 = d_1 = 7 \frac{\lambda_1 D}{d} \quad \text{and} \quad 7\beta_2 = d_2 = 7 \frac{\lambda_2 D}{d}$$

$$\therefore \frac{d_1}{d_2} = \frac{\lambda_1}{\lambda_2}$$

14.

Given,  $\beta = 1 \text{ mm} = (D\lambda/d)$

Distance of 1st bright fringe from the centre,

$$x_n = 2n \left( \frac{D\lambda}{2d} \right)$$

For first bright fringe,  $n = 1$

$$\text{So,} \quad x_1 = 2 \left( \frac{D\lambda}{2d} \right) = \frac{D\lambda}{d} = 1 \text{ mm.}$$

15.

$\lambda = 5000 \text{ \AA}$ ,  $d = 0.2 \text{ mm}$  and  $D = 200 \text{ cm}$

$$x_n = 2n \left( \frac{D\lambda}{2d} \right)$$

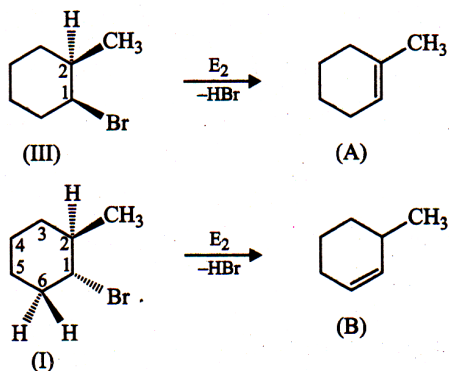
$$\therefore x_3 = 2 \times 3 \left( \frac{D\lambda}{2d} \right)$$

$$= \frac{3 \times 200 \times 5000 \times 10^{-8}}{0.2 \times 10^{-1}} \text{ cm} = 1.5 \text{ cm.}$$

### [CHEMISTRY]

24. (a) : For the same aryl group, *b.p.* increases as the size of the halogen increases. Thus,  $\text{C}_6\text{H}_5\text{I}$  has the highest *b.p.*
25. (c) : Boiling point decreases with branching. Therefore, option (c) is wrong.
26. (d) : Being strained cyclopropane ring readily opens up to form only *n*-propyl bromide. In contrast, reaction (a) gives a mixture of *n*-propyl and isopropyl bromides, reaction (b) gives isopropyl bromide while reaction (c) does not occur at all.
27. (d) : As the size of the alkyl group increases, the  $\text{S}_{\text{N}}2$  reactivity decreases. Further, C-Cl bond is stronger and more difficult to cleave than C-Br bond. Thus, option (d) is correct.
28. (d) : 3-Methyl-3-bromohexane is a  $3^\circ$  alkyl halide and hence undergoes solvolysis (nucleophilic substitution reaction with the solvent) by  $\text{S}_{\text{N}}1$  mechanism. Since  $\text{S}_{\text{N}}1$  reactions do not involve inversion, therefore, option (d) is *incorrect* while all other options are correct.

29. (b) : During the reaction between the optically active alcohol and *p*-toluenesulphonic acid, the C–O bond to the chiral centre is not broken. Instead O–H bond is broken. As a result the configuration of the alcohol is retained in the tosylate (A). However, when tosylate (A) is
30. (b) : In  $E_2$  reactions, *trans*-elimination occurs. Since in compound (III), there is a *trans*-H-atom on  $C_2$  carbon carrying the  $CH_3$  group, therefore elimination occurs readily to give stable alkene (A).



In compound (I), *trans*-H is not available on  $C_2$  but there is a *trans*-H available on  $C_6$ , therefore, elimination occurs on the other side to give less stable alkene (B). Compound (II), however, does not have a *trans*-H on either side (i.e.,  $C_2$  or  $C_6$ ), therefore,  $E_2$  reaction does not occur. Thus, option (b) is correct.